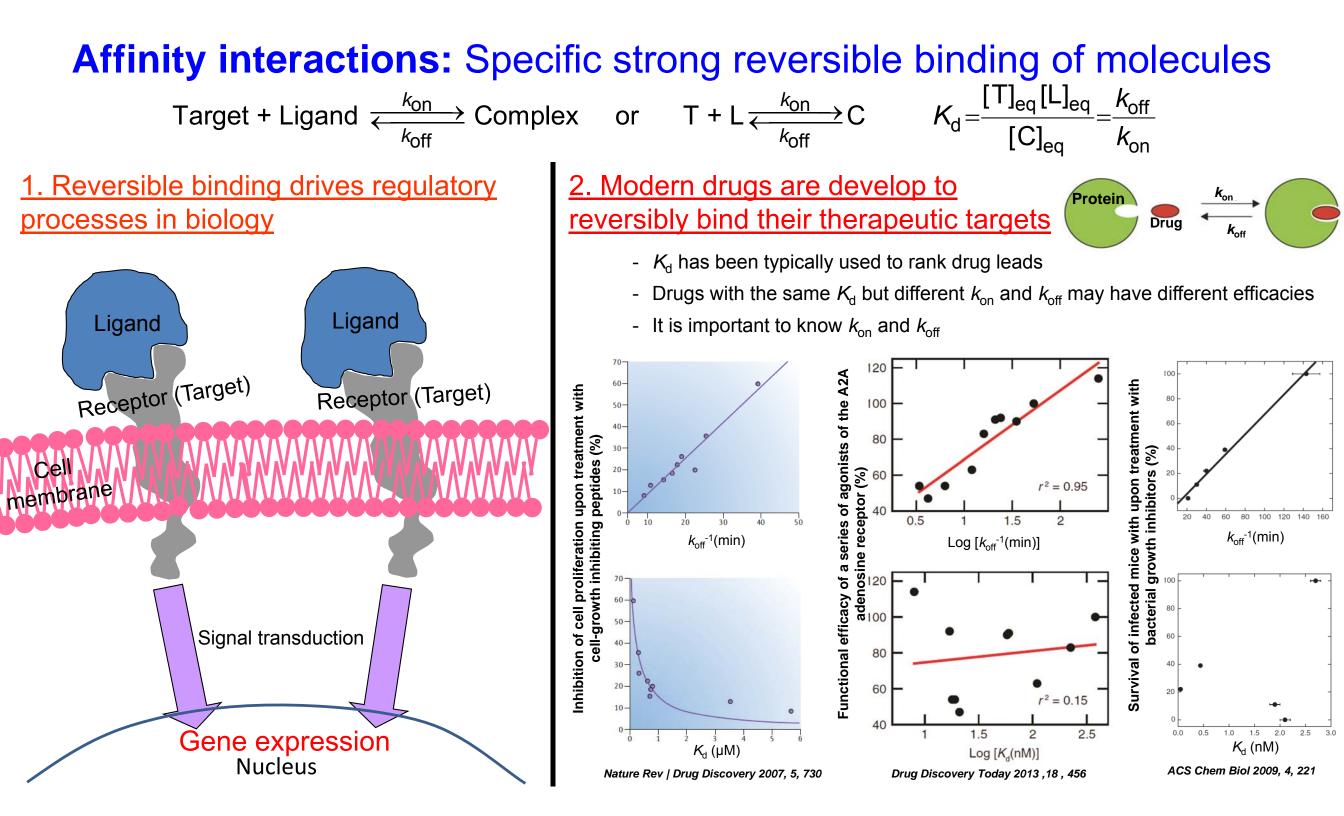
Kinetic Separation: a conceptual platform for development of <u>solution-based kinetic affinity methods</u> (an Analytical Swiss Army Knife)



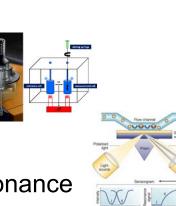
Part 1 Affinity methods and Kinetic Separation in their context



Affinity methods

Finding K_d, k_{on}, k_{off}

- Calorimetry (K_d only)



- Optical methods:
 - surface plasmon resonance
 - biolayer interferometry

Quantitative detection using affinity probes

- Immunoassays (mainly for proteins)
- Hybridization assays (for DNA and RNA)

Selection of affinity probes and drug candidates from complex mixtures

- Affinity chromatography
- Filtration

Limitation of conventional affinity methods:

1. Different concepts 2. Different instruments 3. Different mathematics

Our goal is to develop a multi-faceted approach which facilitates three applications:

- Finding k_{on} , k_{off} , K_{d}
- Quantitative detection
- Selection of affinity probes and drug candidates

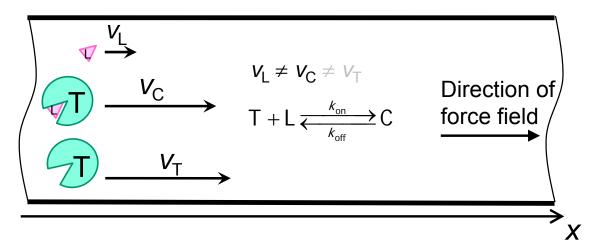
based on:

1. Single concept 2. Single instrument 3. Single mathematics



Our approach: Kinetic Separation

<u>Definition:</u> Kinetic Separation is separation of interacting species in a narrow tube without immobilization of T or L



Two major processes

- 1. Reversible binding of T to L
- 2. Migration of **T**, **L**, and **C** with different velocities

Mass transfer is described by a system of differential equations:

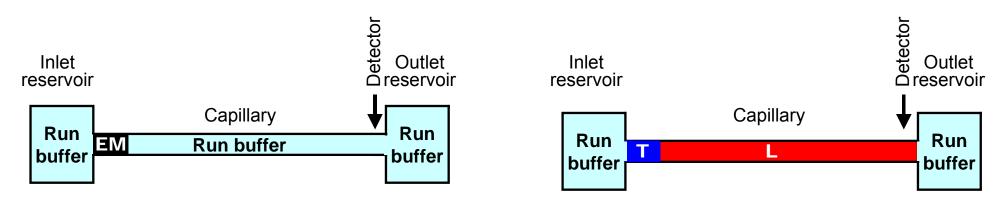
$$\frac{\partial [T]}{\partial t} + \mathbf{V}_{T} \frac{\partial [T]}{\partial x} = -\mathbf{k}_{on}[T][L] + \mathbf{k}_{off}[C]$$
$$\frac{\partial [L]}{\partial t} + \mathbf{V}_{L} \frac{\partial [L]}{\partial x} = -\mathbf{k}_{on}[T][L] + \mathbf{k}_{off}[C]$$
$$\frac{\partial [C]}{\partial t} + \mathbf{V}_{C} \frac{\partial [C]}{\partial x} = -\mathbf{k}_{off}[C] + \mathbf{k}_{on}[T][L]$$

 k_{on} and k_{off} can be found by solving these equations for experimental [L](t) and known velocities

- Qualitatively unique sets of initial and boundary conditions define different methods of Kinetic Separation



Schematic representation of initial and boundary conditions in Kinetic Separation

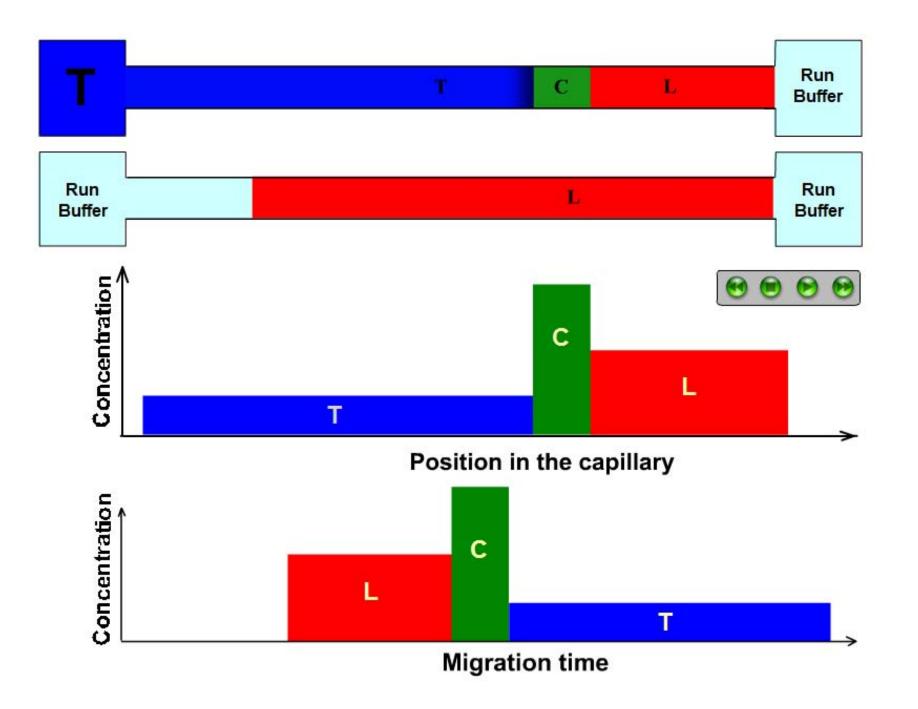












Separation and detection approaches suitable for Kinetic Analysis Separation

- <u>Requirements:</u> 1. Different velocities of L and C (and T): $v_{L} \neq v_{C} \neq v_{T}$
 - 2. Negligible influence of separation on k_{on} and k_{off}

Suitable approaches:

- **1.** Affinity chromatography (affects k_{on} and k_{off})
- 2. Reversed phase chromatography (affects k_{on} and k_{off})
- 3. Ion exchange chromatography (affects k_{on} and k_{off})
- 4. Kinetic Capillary Electrophoresis (KCE)
- 5. Sedimentation (different gravity)
- 6. Kinetic Size-Exclusion Chromatography (KSEC)

Detection

<u>Requirements:</u> 1. Negligible influence of detection on k_{on} and k_{off}

2. nM sensitivity for studying complexes with nM K_{r} values

Suitable approaches:

- 1. Fluorescence labeling (can affect k_{on} and k_{off} of protein-small molecule binding)
- 2. Label-free optical detection (low sensitivity)
- 3. Mass Spectrometry (MS)
- There are 4 practical kinetic separation options:

KCE-Fluor. and KSEC-Fluor. KCE-MS and KSEC-MS

Proven applications of Kinetic Separation

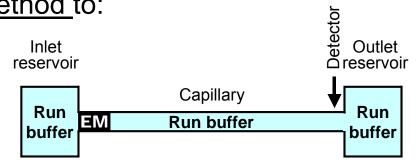
- 1. Finding k_{on} , k_{off} , and K_{d} for protein-ligand binding
- 2. Finding ΔH and ΔS of protein-ligand binding
- 3. Affinity analyses of proteins using DNA aptamers as affinity probes
- 4. Selection of "smart" ligands (ligands with desirable range of k_{off} or K_d)

All developed methods have been fully documented and published or patented for end users

JACS 2013, 134, 8041 JACS 2011, 133, 12486 JACS 2010, 132, 13639 JACS 2010, 132, 7062 JACS 2008, 130, 11862 JACS 2008, 130, 9137 JACS 2007, 129, 7260 JACS 2006, 128, 1410 JACS 2005, 127, 17104 JACS 2005, 127, 11224 JACS 2005, 127, 3165 JACS 2004, 126, 7166 JACS 2003, 125, 13451 JACS 2002, 124, 13674	Anal. Chem. 2015, 87, 3099 Anal. Chem. 2015, 87, 2474 Anal. Chem. 2015, 87, 1411 Anal. Chem. 2015, 87, 1411 Anal. Chem. 2014, 86, 10016 Anal. Chem. 2014, 86, 1298 Anal. Chem. 2012, 84, 6944 Anal. Chem. 2011, 83, 8617 Anal. Chem. 2011, 83, 8387 Anal. Chem. 2011, 83, 7582 Anal. Chem. 2011, 83, 6330 Anal. Chem. 2011, 83, 1381 Anal. Chem. 2010, 82, 8692 Anal. Chem. 2010, 82, 8637	Anal. Chem. 2010, 82, 4428 Anal. Chem. 2009, 81, 490 Anal. Chem. 2007, 79, 1097 Anal. Chem. 2006, 78, 4803 Anal. Chem. 2006, 78, 3171 Anal. Chem. 2006, 78, 2035 Anal. Chem. 2006, 77, 1526 Anal. Chem. 2004, 76, 7114 Anal. Chem. 2004, 76, 1507 Anal. Chem. 2003, 75, 1382	Analyst 2015, 140, 2797 Analyst 2015, 140, 990 Anal. Chim. Acta 2010, 681, 92 Nucl. Acid. Res. 2009, 37, e62 Anal. Chim. Acta 2009, 631, 102 Nature Protocols 2006, 1, 1359 Mol. Cell. Biol. 2007,27, 20 Electrophoresis 2007, 28, 69 Biochemistry 2006, 45, 6075 J. Biomol. Screen. 2006, 11, 115 Anal. Chim. Acta 2006, 564, 91 FEBS Lett. 2005, 579, 1371 Analyst 2003, 128, 571	U.S. patent No 7,666,660, Feb 23, 2010 U.S. patent No 7,672,786, Mar 2, 2010 U.S. patent No 8,224,582, Jul 17, 2012
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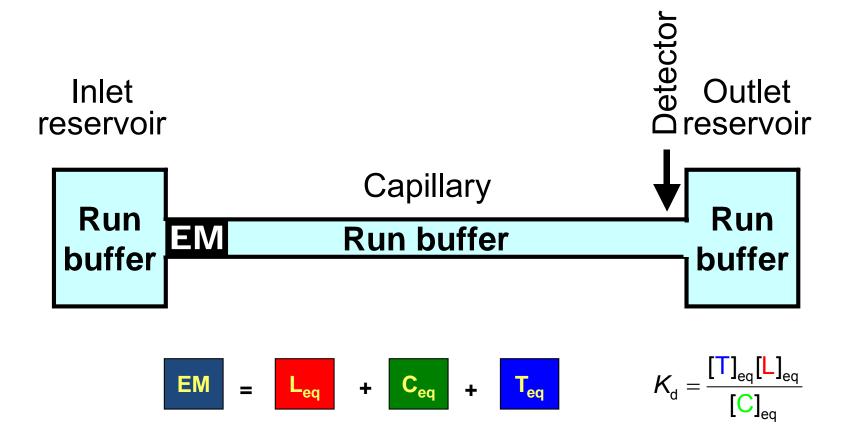
Subject of this lecture is application of a single Kinetic Separation method to:

- (i) selection of ligands from combinatorial libraries
- (ii) kinetic characterization of target-ligand binding
- (iii) calibration-free affinity analysis of target using ligand as affinity probe

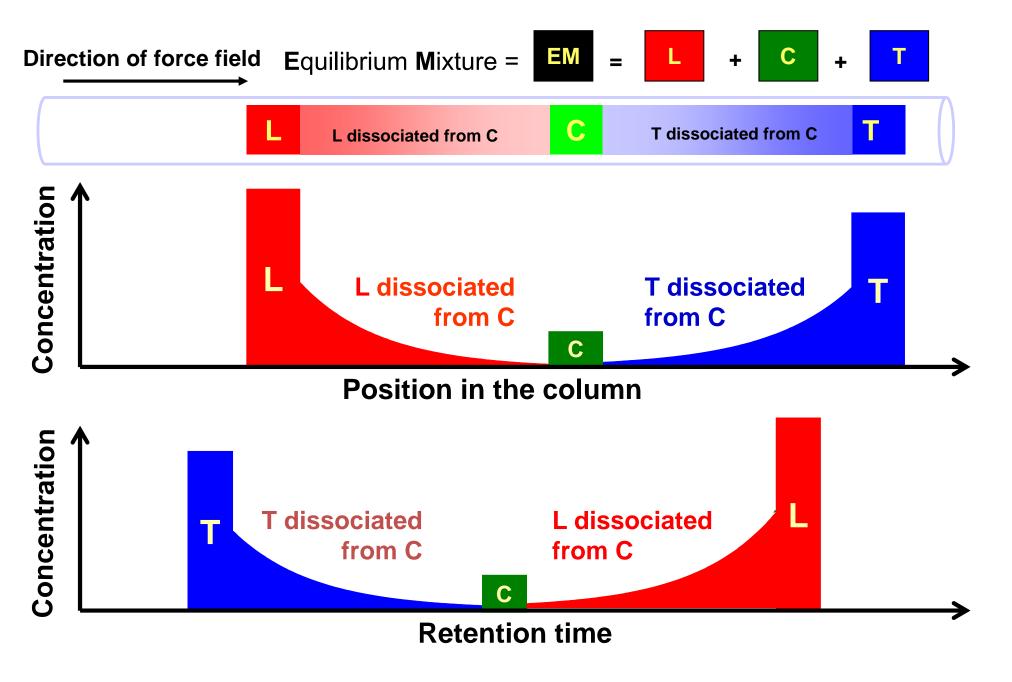


Part 2

Conceptual explanation of one Kinetic Separation method and its applications



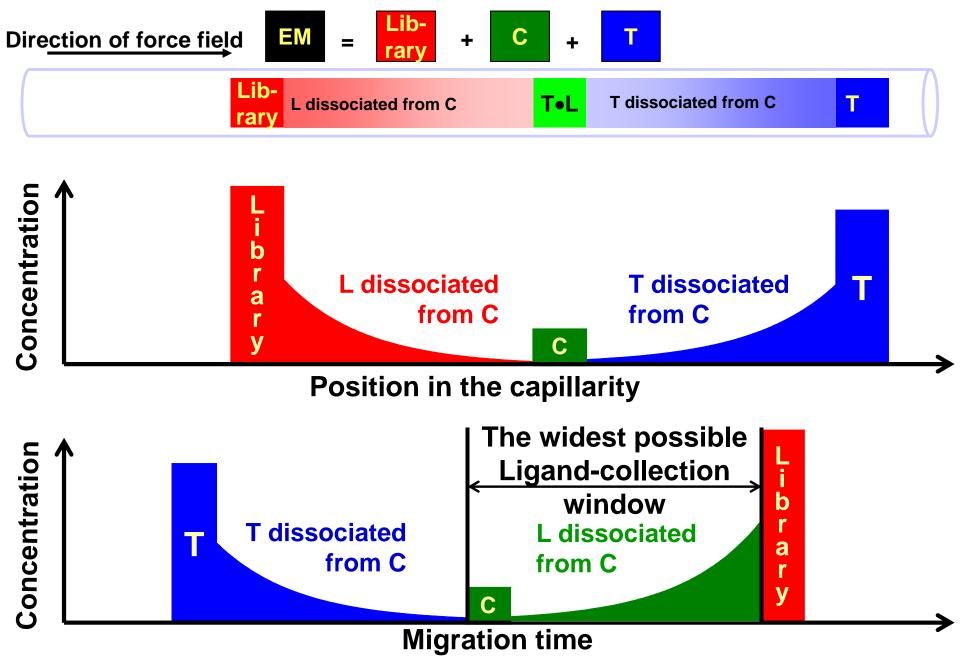
Migration of zones in Kinetic Separation



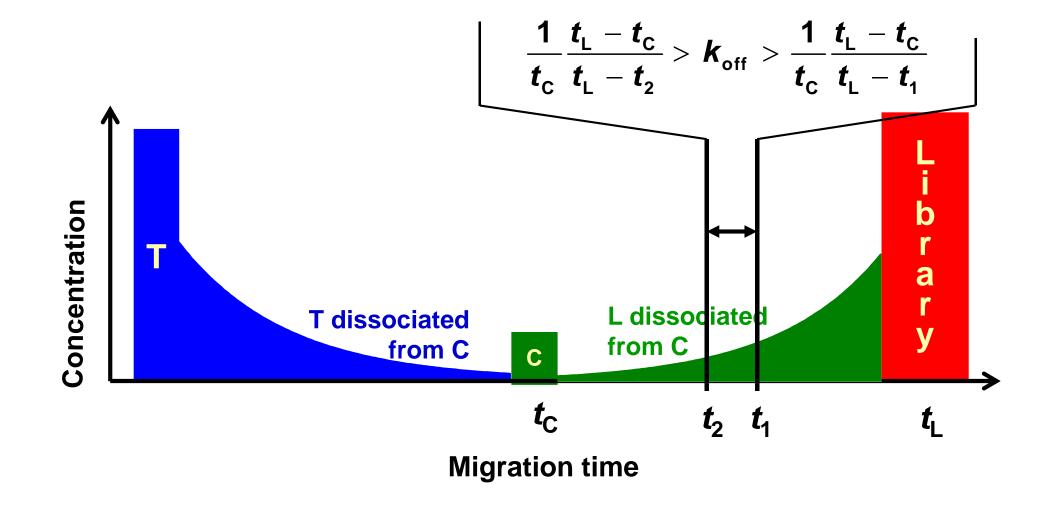
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Concept of Kinetic-Separation-based selection of naive ligands





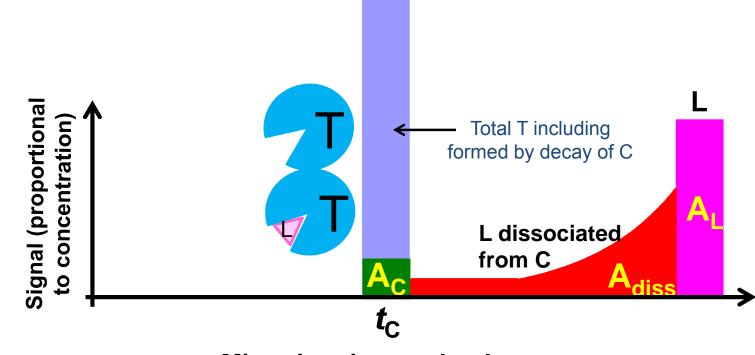
Concept of Kinetic-Separation-based selection of smart ligands with desirable k_{off}



Concept of Kinetic-Separation-based (i) determination of k_{off} and k_{on} and (ii) calibration-free quantitative analysis of T **Case 1:** the 3 zones are separated \Rightarrow we can assume that there is no rebinding of T and L formed by the decay of C Signal (proportional to concentration) Α T is not spectroscopically L dissociated from C visible **t**l Migration time to the detector $K_{d} = \frac{\left[T\right]_{0} \left(1 + A_{L} / (A_{C} + A_{diss})\right) - \left[L\right]_{0}}{1 + (A_{C} + A_{diss}) / A_{L}}$ $k_{\rm off} = \ln$ 1) / *t*_c, $k_{\rm on} = k_{\rm off} / K_{\rm d}$ $\frac{\overline{A_{diss}} + A_{C}}{1 + A_{L}} + [L]_{0} \frac{1 + A_{L}}{1 + A_{L}} \frac{1}{A_{L}}$ $[\mathsf{T}]_0 = K_d$ 2)

Concept of Kinetic Separation-based (i) determination of k_{off} and k_{on} and (ii) calibration-free quantitative analysis of T

<u>**Case 2</u>**: T is not separated from C \Rightarrow the presence of T in the zone of C results in re-binding of L (formed from the decay of C) to the excess of T</u>

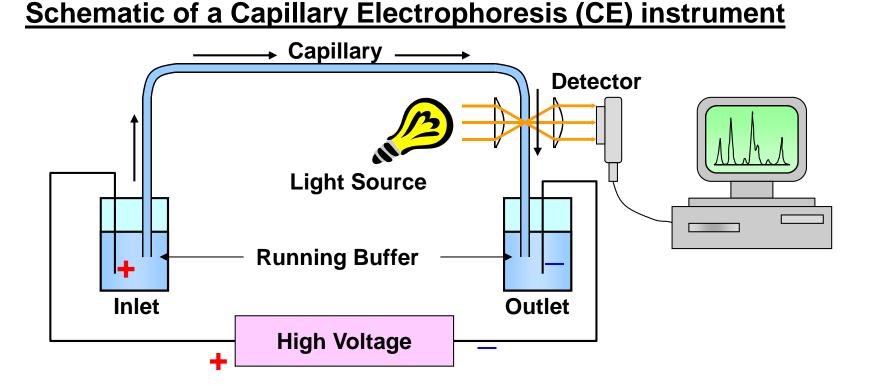


Migration time to the detector

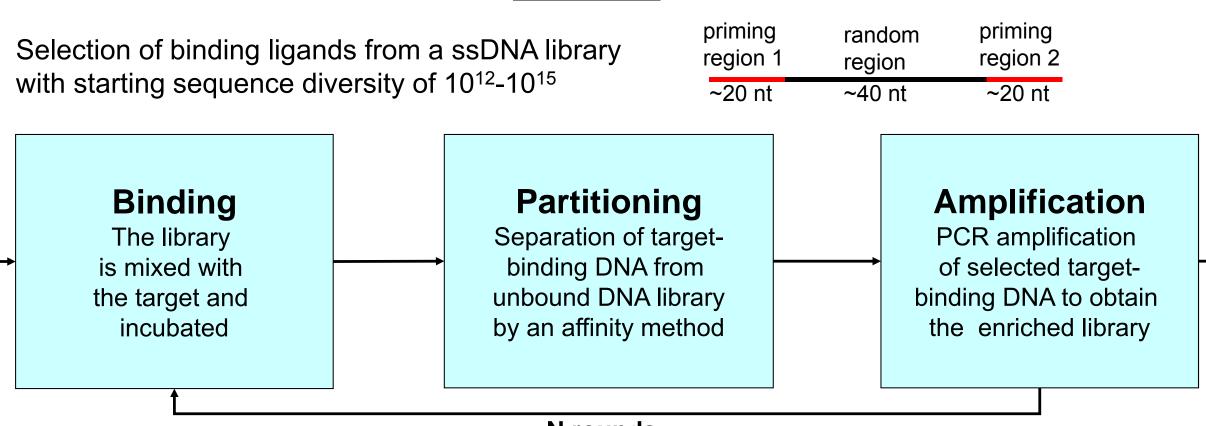
- No simplifying assumptions can be made for finding k_{on} and k_{off} or $[T]_0$
- Fitting an experimental concentration profile by a simulated one is required to find k_{on} and k_{off} or find [T]₀ for known k_{on} and k_{off}

Part 3

Examples of applications of <u>Kinetic Capillary</u> <u>Electrophoresis</u> (KCE) with optical detection



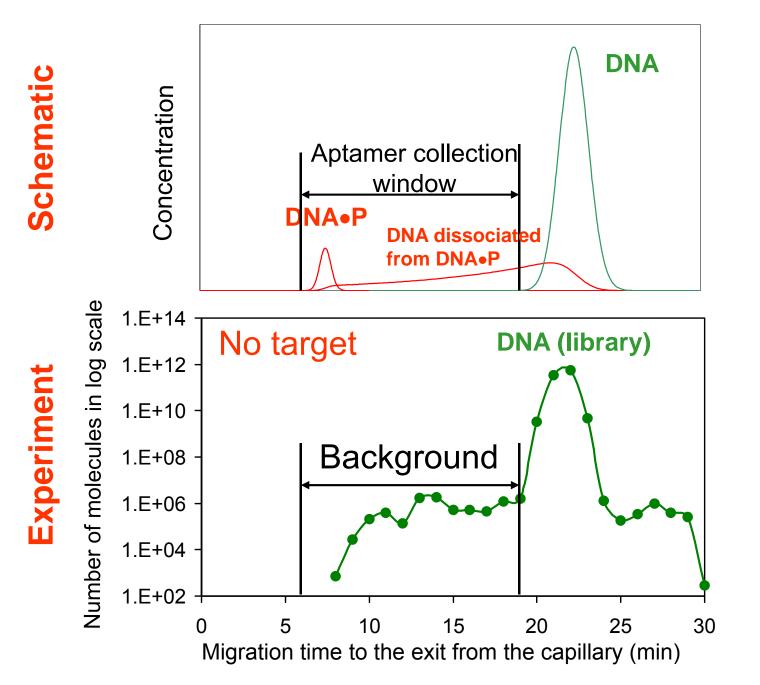
Application 1 Selection of DNA aptamers by KCE SELEX



N rounds

- Conventional SELEX uses partitioning on surfaces, *e.g.* on filters
- Background of partitioning is typically 1-10%
- Required number of rounds of selection is typically more than 10
- Smart aptamers (e.g. with pre-defined k_{off} or K_d) are hard to obtain

Advantage of KCE-based partitioning is exceptionally low background



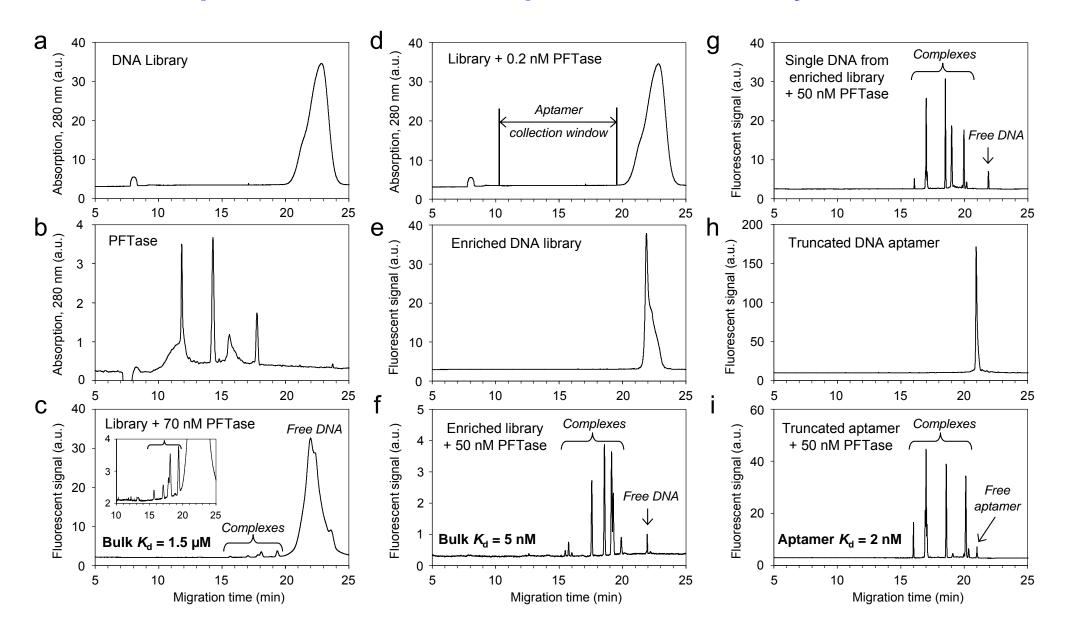
Background is the relative amount of DNA in the aptamer collection window without target protein

For KCE: Background < 0.001%

For other methods: Background > 0.1%

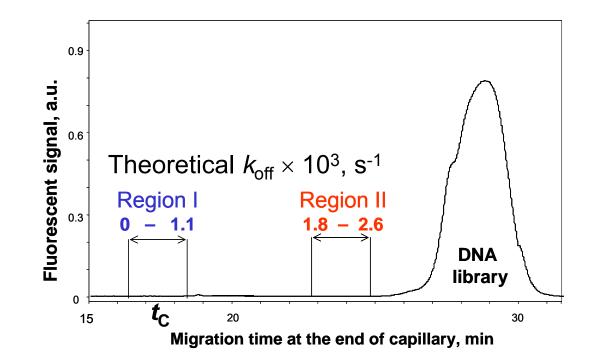
Example 1:

One-step selection of naive aptamers to farnesyltransferase



Example 2:

Selection of smart aptamers with pre-defined k_{off} for MutS protein



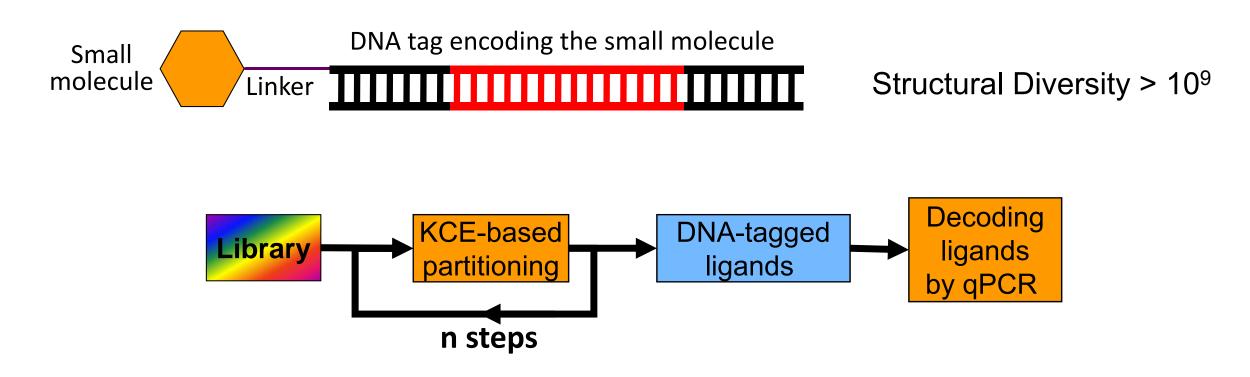
	k _{off}	k _{off}	K _d
	(theoretical), $x10^{-3} s^{-1}$	(experimental), $x10^{-3} s^{-1}$	(experimental),
	$x10^{-3} s^{-1}$	$x10^{-3} s^{-1}$	nM
Region I	0-1.05	0.4	11
Region II	1.76-2.64	1.7	44

Application 2

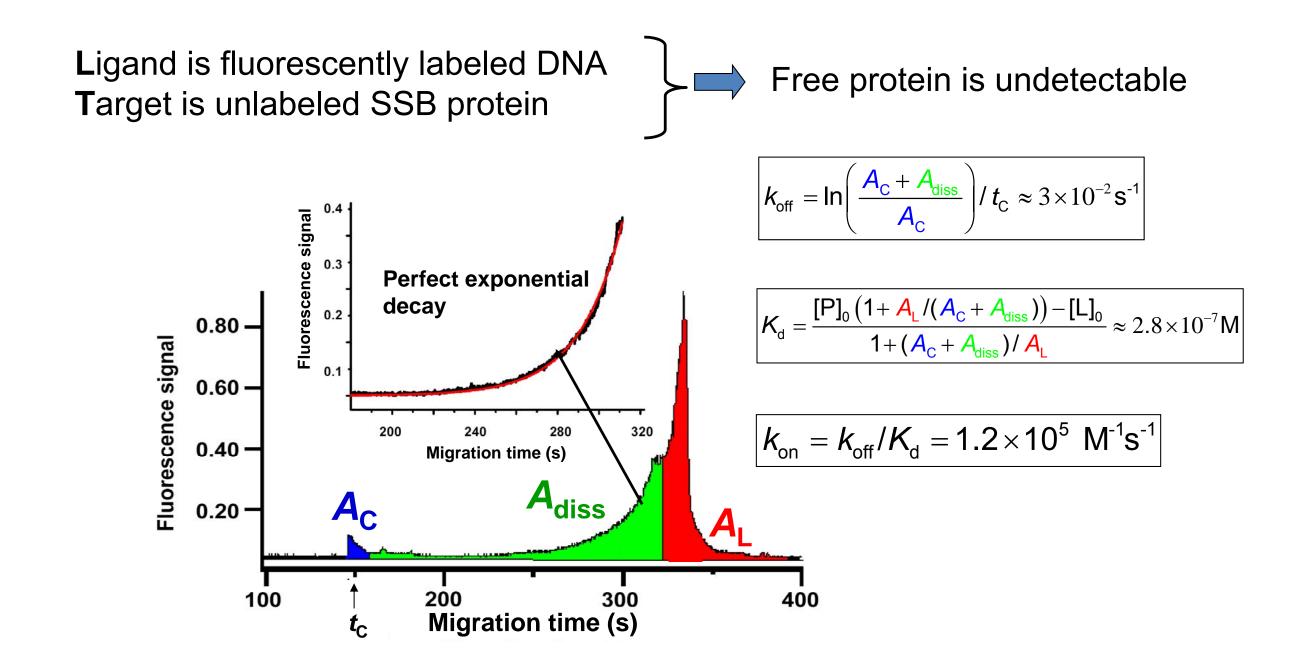
KCE-based selection of drug leads from libraries of DNA-encoded small molecules

(current work in collaboration with GlaxoSmithKline)

High efficiency of KCE-based partitioning is attractive for selection of binding ligands from libraries of DNA-encoded small molecules



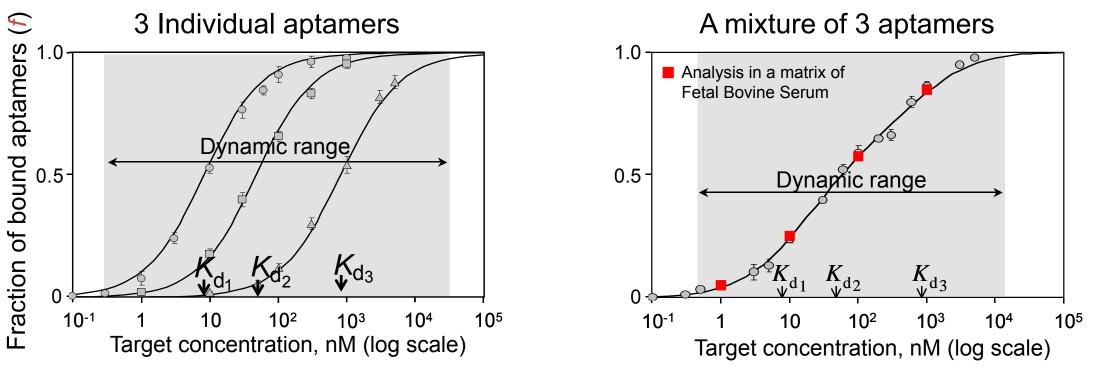
Application 3 KCE-based determination of and k_{off} and k_{on} of protein-DNA binding



Application 4

KCE-based calibration-free analysis of MutS protein with ultra-wide dynamic range

3 Smart Aptamers with K_d of 7.6, 46, and 810 nM were used



 $[T]_0$ is found by solving the following algebraic equation for *n* aptamers (L):

$$\sum_{i=1}^{n} \frac{[L]_{0_{i}}}{K_{d_{i}} + [T]_{0} - f \times \sum_{j=1}^{n} [L]_{0_{j}}} = \frac{f \times \sum_{i=1}^{n} [L]_{0_{i}}}{[T]_{0} - f \times \sum_{i=1}^{n} [L]_{0_{i}}}$$

Part 4

Examples of applications of Kinetic Size-Exclusion Chromatography (KSEC) with MS detection

Instrumentation for KSEC-MS

Separation of L from C on a SEC column by HPLC

MS detection of signal proportional to [L]

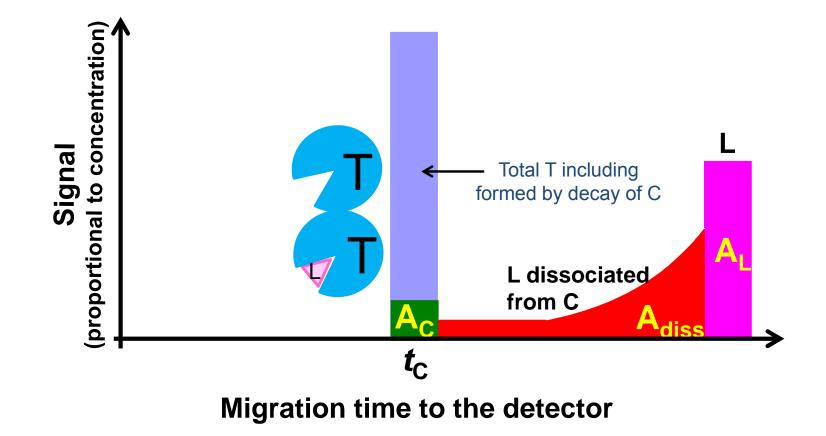






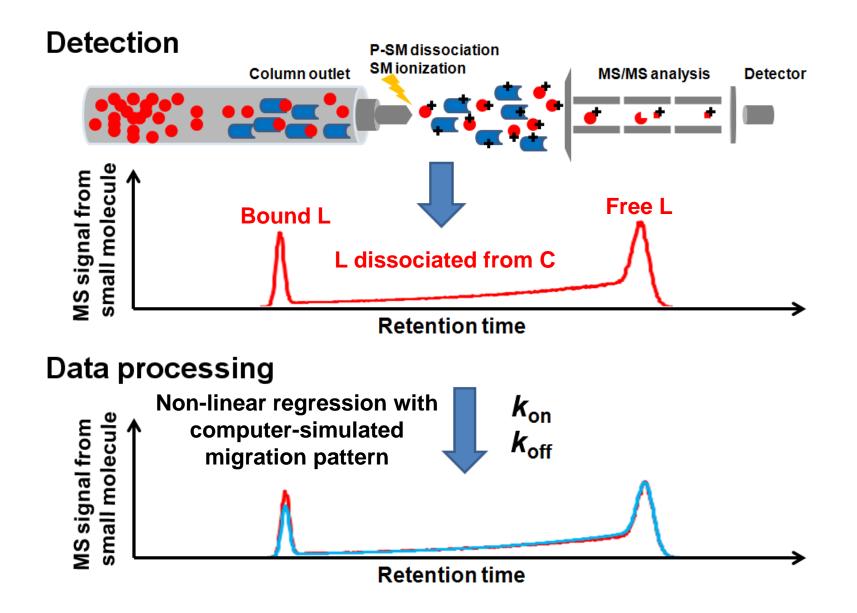
Application 1

Determination of k_{on} and k_{off} for binding of a protein to a small-molecule drug



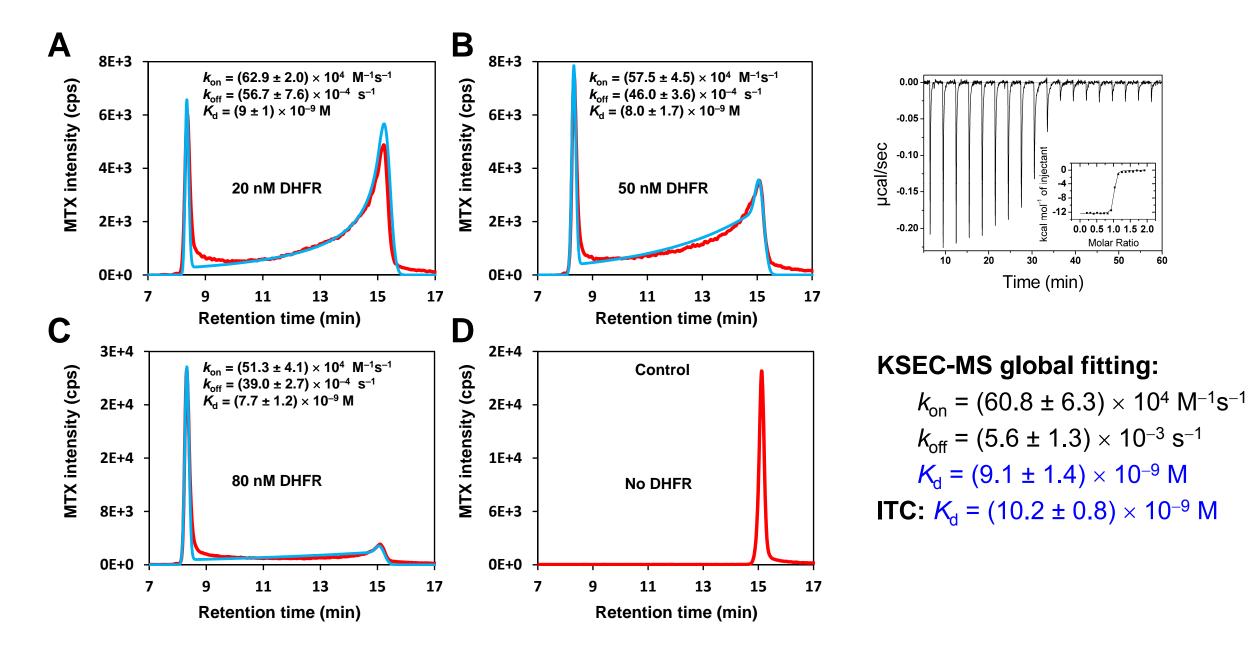
Fitting an experimental signal from L by a simulated one is required to find k_{on} and k_{off}

Small molecule detection and determination of kon and koff



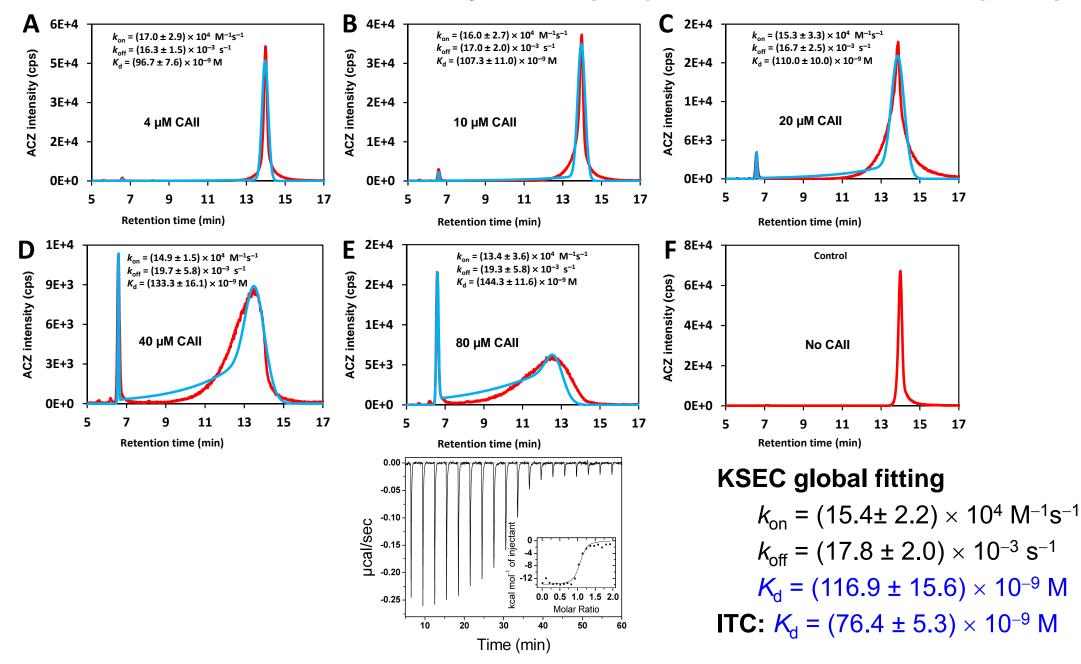
Example 1:

Interaction between dihydrofolate reductase (DHFR) and Methothrixate (MTX)



Example 2

Interaction between carbonic anhydrase (CA) and acetazolamide (ACZ)



Conclusions

Kinetic Separation is an "Analytical Swiss Army Knife"

Science is always wrong. It never solves a problem without creating ten more!

George Bernard Shaw

More problems

- 1. Interfacing CE with MS for **Kinetic Separation**
- 2. Development of Kinetic Separation tools for studying protein-protein binding
- 3. Development of **Kinetic Separation** tools for studying binding stoichiometry
- 4. Development of **Kinetic Separation** for studying kinetics of assembly/disassembly of complex molecular machines
- 5. Finding solutions for **Kinetic Separation** with physiological run buffers
- 6. Expanding **Kinetic Separation** to new other separation modes (*e.g.* ultracentrifugation)



